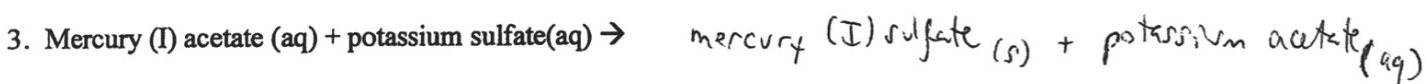
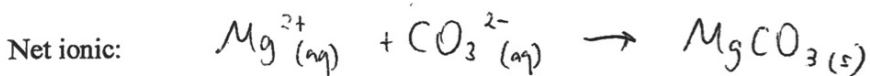
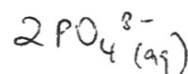
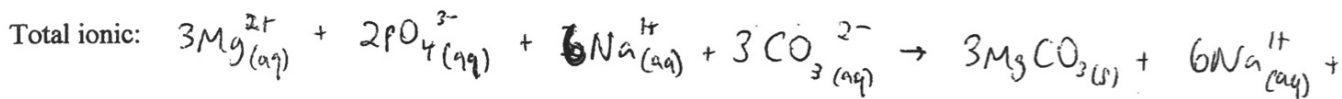
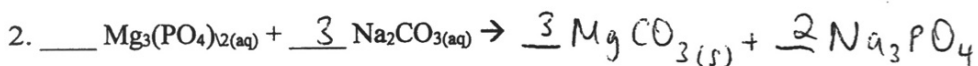
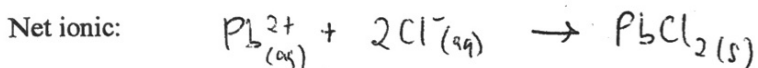
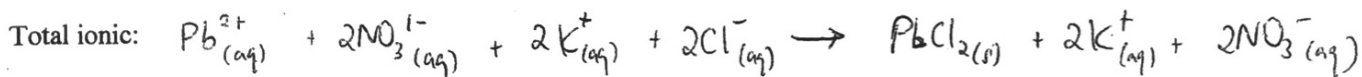
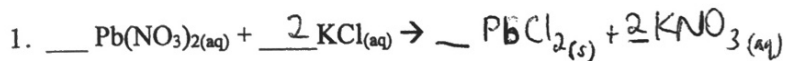
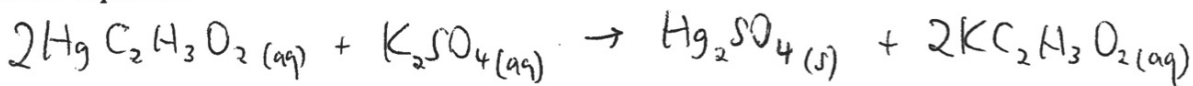


Name KEYPeriod **Net Ionic Equations: More Practice Problems**

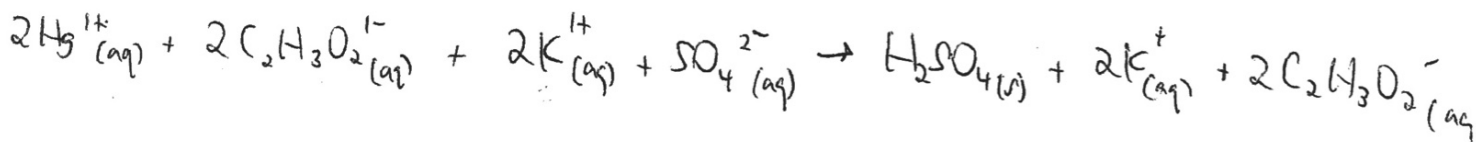
Complete and balance the molecular equations for the following sets of reactants. Then, write the total ionic and net ionic forms of the equations. If all species are spectator ions, please indicate that no reaction takes place. Be sure to use your solubility chart.



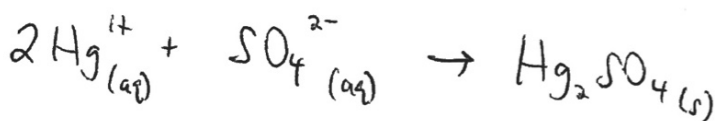
Total Molecular Equation:



Total ionic:

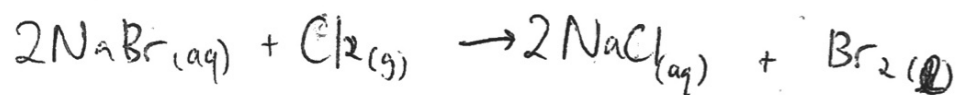


Net ionic:

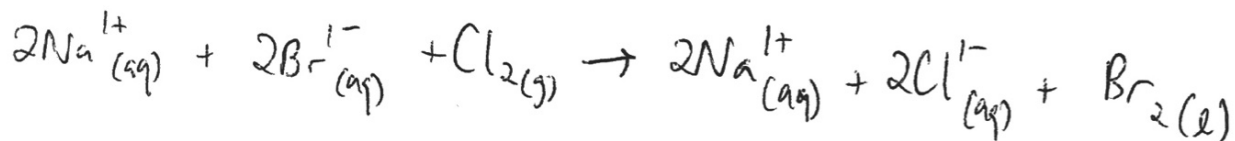


4. Aqueous sodium bromide and chlorine gas produce... sodium chloride and bromine

Total Molecular Equation:



Total ionic:



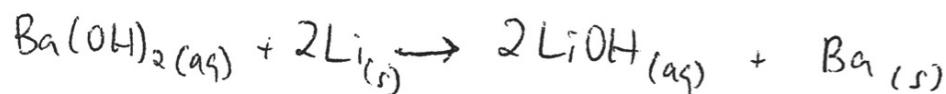
Net ionic:



5. Barium hydroxide and lithium produce... lithium hydroxide and barium

↑
assume aqueous

Total Molecular Equation:



Total ionic:



Net ionic:

